Overview

- Learning and simulations
- Research questions
 - Diagnosing learning behaviour
 - Curriculum planning
 - Explanation (incl. visualisation) (*brief*)
- Concluding remarks
- Applications and References (*brief*)

Cognitive diagnosis: Outstanding problem in AIED

Goal:

interpretation of the 'problem solving behaviour' of a learner (e.g. correct, incorrect, incorrect in a certain way, incomplete, etc.)

That is:

construct a model of the knowledge state of a learner

In order to:

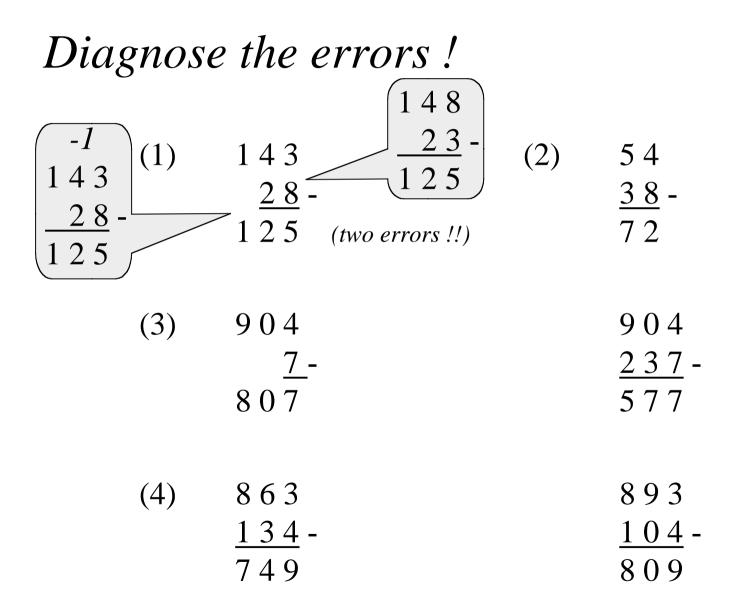
determine next step in the interaction

(e.g. Explanation / Re-mediation / Assignment / etc.)

Useful to distinguish between

- local (w.r.t. a specific answer)
- global (learner behaviour over time)

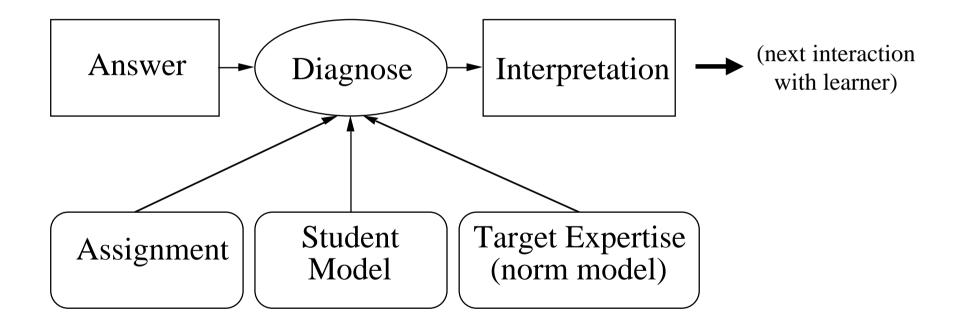
Cognitive diagnosis: subtraction, a typical research domain



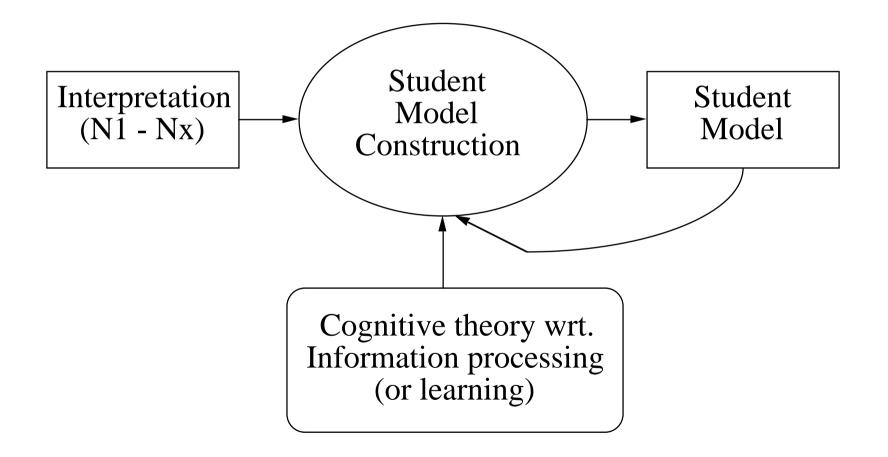
Cognitive diagnosis: *Typical approaches & their problems*

- Overlay
 - novices behave different from experts
 - no knowledge of faults
- Catalogue of Bugs / Misconceptions
 - a lot of work
 - domain dependent
 - never complete
- Generative
 - requires theory of cognition/learning
 - requires domain specific pruning
- Model-Based Approach ?! (= consistency-based diagnosis)

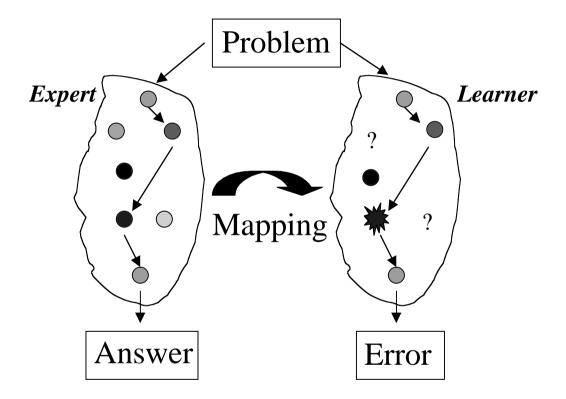
Cognitive diagnosis: Local view



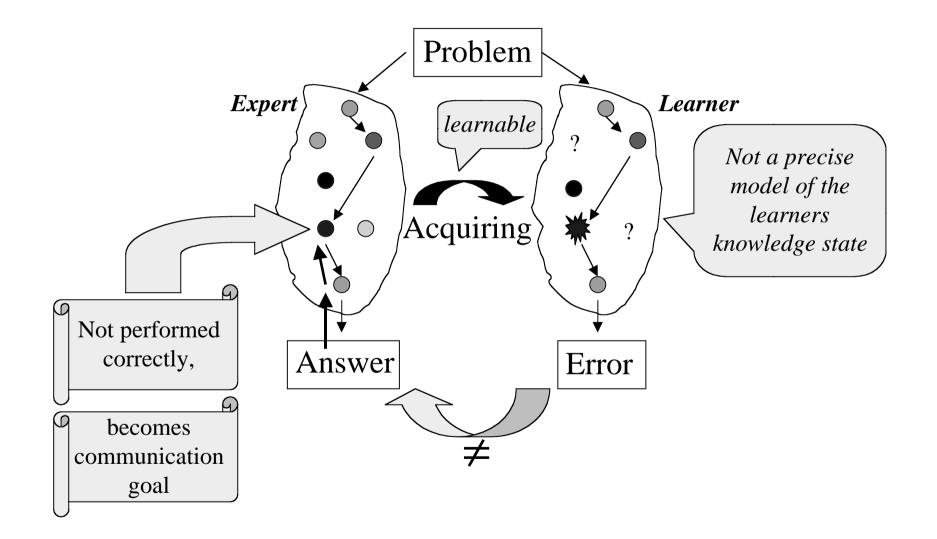
Cognitive diagnosis: Global view



Cognitive diagnosis: previous techniques



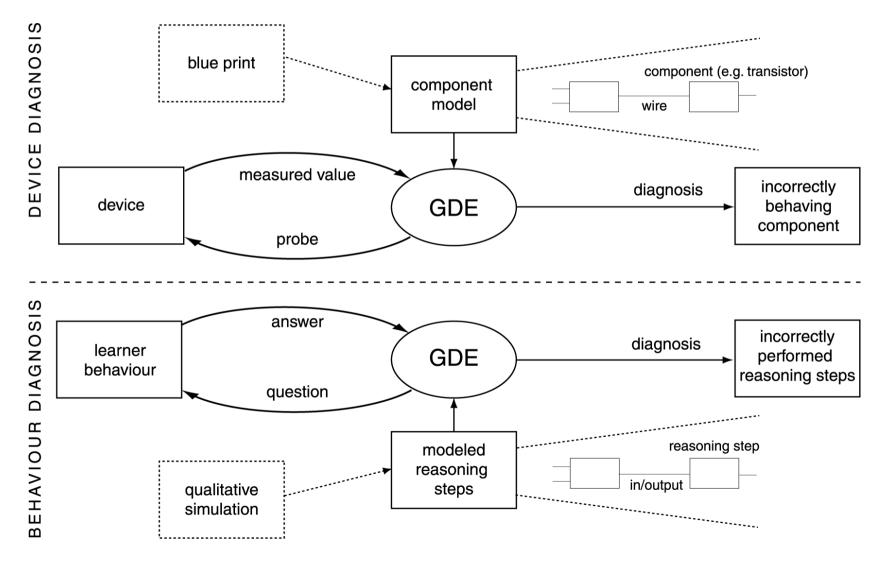
Cognitive diagnosis: consistency based



(Inspired on experiments with real teachers, see: de Koning, 1998)

GDE for Diagnosing Learner Behaviour

Basic idea



GDE for Diagnosing Learner Behaviour

What to do? (de Koning et al., 2000)

Base Model

- Cognitive validity (inference steps) (see: de Koning, 1998)
- Component-connection paradigm (Qualitative Model → GDE)

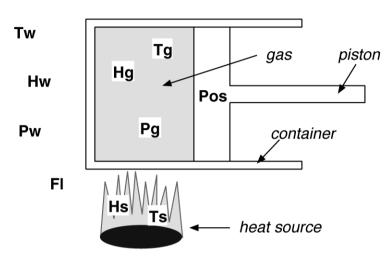
Make it work

- Diagnostic engine (→ aggregation..)
- Probe selection & Probing

Evaluate

- Diagnosis
- Approach as a whole

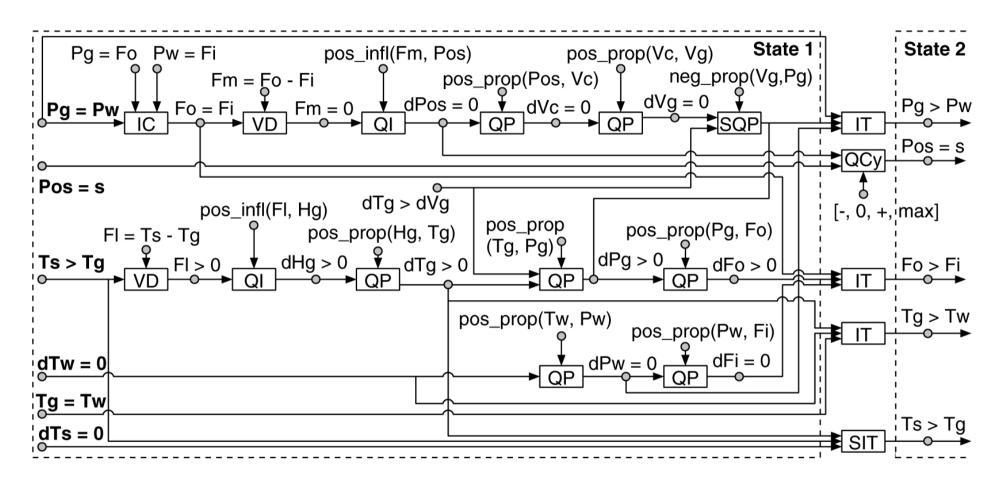
Example: The Container-Piston System



- = temperature of the surrounding world Tw = heat of the surrounding world Hw = pressure of the surrounding world Pw = temperature of the gas Тα **Hg** = heat of the gas = pressure of the gas Pa = temperature of the heat source Ts = heat of the heat source Hs = heat flow between heat source and gas FL
- **Pos** = position of the piston

"Because the temperature of the heater is higher than the temperature of the gas, there will be a heat flow from the heater to the gas. Therefore, the temperature of the gas will no longer be equal to the temperature of the outside world, but will become higher."

The Base Model

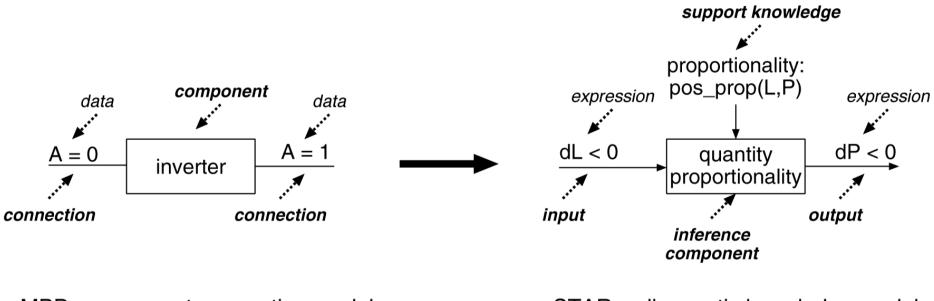


- IC = inequality correspondence
- VD = value determination
- QI = quantity influence
- QP = quantity proportionality

- SQP = submissive quantity proportionality
 - IT = inequality termination
 - SIT = submissive inequality termination
- QCy = quantity continuity

The Base Model *issues* (1*a*)

Components and Component Behaviour



MBD: component-connection model

STAR: diagnostic knowledge model (base model)

The Base Model *issues* (1b)

Components and Component Behaviour

type: Quantity Influence

ports:

 $\underline{In} =$ quantity value, $\underline{Sup} =$ influence, $\underline{Out} =$ quantity derivative

example:

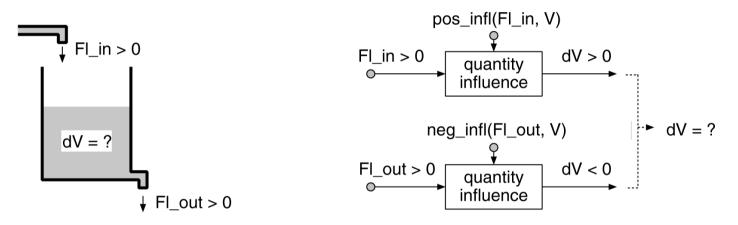
'There is a [positive] flow, so the volume decreases' ($Fl > 0 \& neg_infl(Fl,V) \rightarrow \partial V < 0$)

component description (behaviour rules):

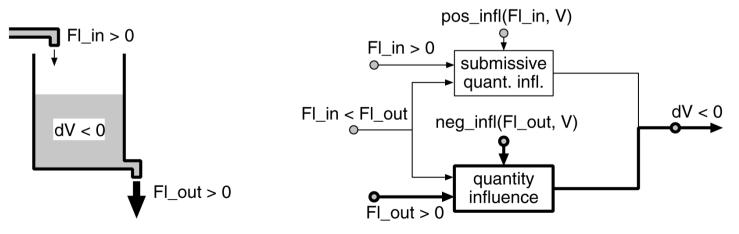
Forward propagation: In & Sup \rightarrow Out IF In = [A = >/=/< = 0] & Sub = [$pos_infl(A,B)$] THEN Out = [$\partial B = +/0/-$] IF In = [A = >/=/< = 0] & Sub = [$neg_infl(A,B)$] THEN Out = [$\partial B = -/0/+$] Backward propagation: Out & Sub \rightarrow In In $\neq 0$ & Out $\neq 0 \rightarrow$ Sub

The Base Model *issues* (2)

Multiple influences / proportionality's / terminations Have <u>context</u> <u>dependent</u> behaviours...



Solution: submissive components (e.g. submissive influence)



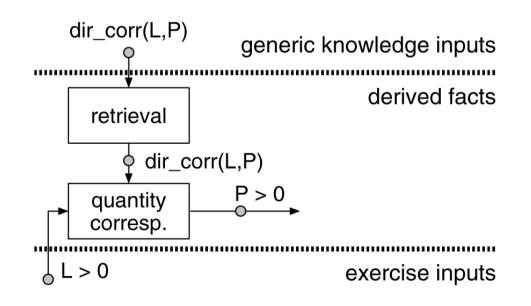
The Base Model *issues* (3)

Only components can become a diagnosis

How to distinguish between

- knowing what dependencies exist and apply
- how to reason with them (and other input / outputs)

Solution: introduction of <u>retrieval</u> components



GDE for Diagnosing Learner Behaviour

First try... It didn't work

Too many single faults...

- qualitative calculus (weak: ambiguous solutions)
- too few connections (fewer constraints)
- missing backwards propagation rule (a minor problem)

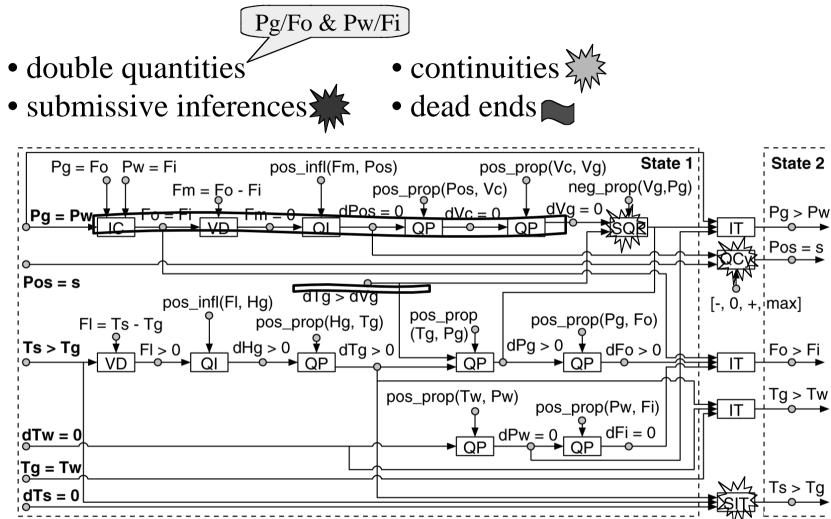
Solution...

Use hierarchy of models, but ...

no hierarchy information available...

→ therefore: exploit component types

Hiding of Inessential Details

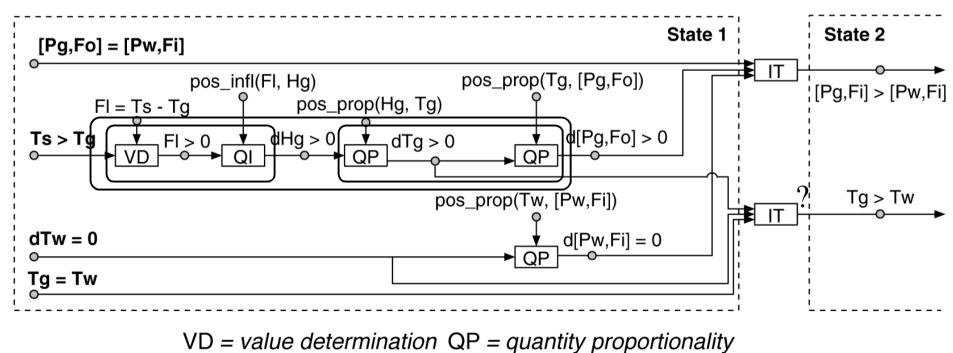


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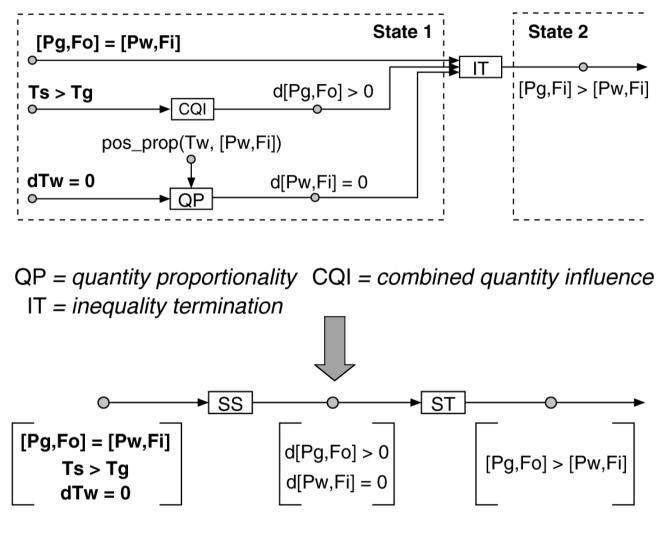
- SQP = submissive quantity proportionality
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 - SIT = submissive inequality termination
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Chunking

- transitive chunking
- pred. key component chunking
- succ. key component chunking



Grouping



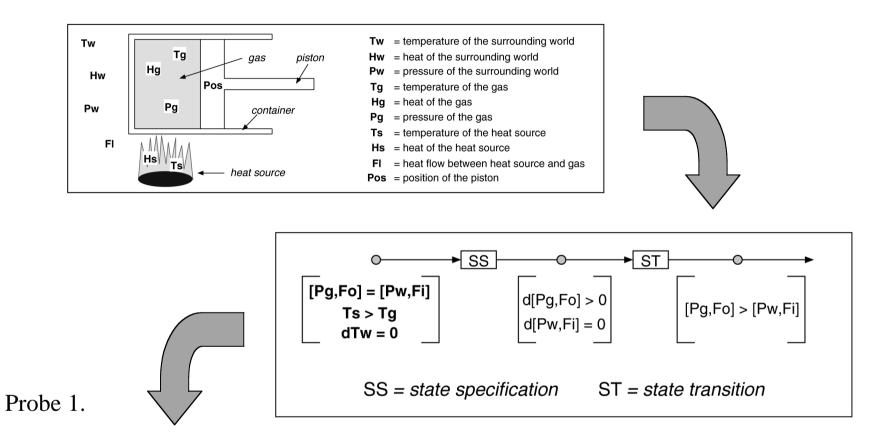
SS = state specification ST = state transition

Container-piston example: Results

Number of <u>inference</u> components after different aggregation techniques

technique	nr. of
used	components
(base model)	824
hiding	403
chunking	220
grouping	16

An Example Diagnosis (1) (General question / Probe 1)



Teacher: The pressure of the gas is initially equal to the pressure of the outside world
(Pg = Pw). What do you think about this pressure ratio in the next behavioural state?a. Pg < Pw
b. Pg = Pw
c. Pg > PwLearner: b. Pg = Pw
(NB: incorrect answer)

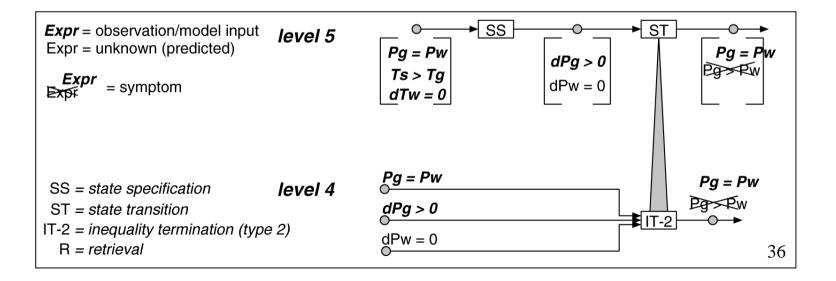
An Example Diagnosis (2) (probe 2)

• Two expressions can be measured: $\partial Pg > 0$ or $\partial Pw = 0$. (Suppose we ask the first one.)

Teacher:Is the pressure of the gas initially:
a. increasing
b. steady
c. decreasingLearner:a. increasing.(NB: correct answer)

• Diagnosis: Single fault diagnosis ST

• Decompose: ST (but, empty conflict set at next level)



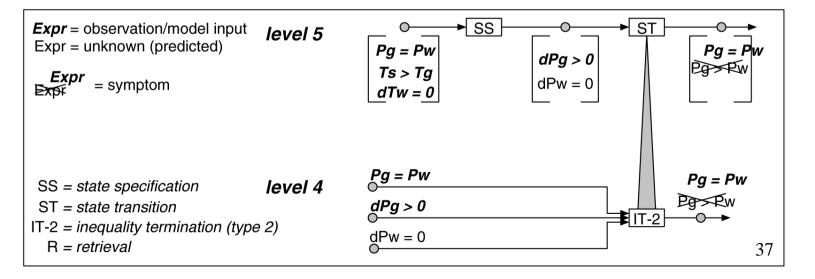
An Example Diagnosis (3) (probe 3)

• The only resulting expression to ask for is $\partial Pw = 0$. (input for ST)

Teacher: Is the pressure of the world initially: a. increasing b. steady c. decreasing Learner: b. steady. (*NB: correct answer*)

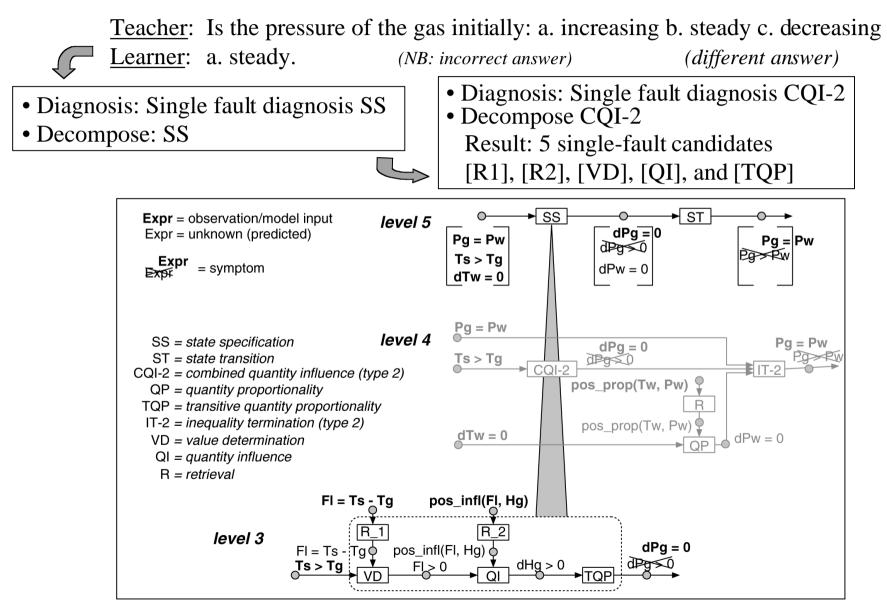
- Diagnosis: Single fault diagnosis ST
- Decompose: ST

• Diagnosis: Single fault diagnosis IT-2 (= diagnosed fault)



An Example Diagnosis (4) (probe 2b)

• Two expressions can be measured: $\partial Pg > 0$ or $\partial Pw = 0$. (Suppose we ask the first one.)



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An Example Diagnosis (5) (probe 3b)

- Five components can be tested: [R1], [R2], [VD], [QI], or [TQP]
- Probe selection:
 - probabilities: 1.43 for [R1] and [R2]; 2.0 for [TQP]; 1.0 for [VD] and [QI]
 - best measurement Fl > 0 because: |(1.43 + 1) (1.43 + 1 + 2)| = 2 (*best split*)

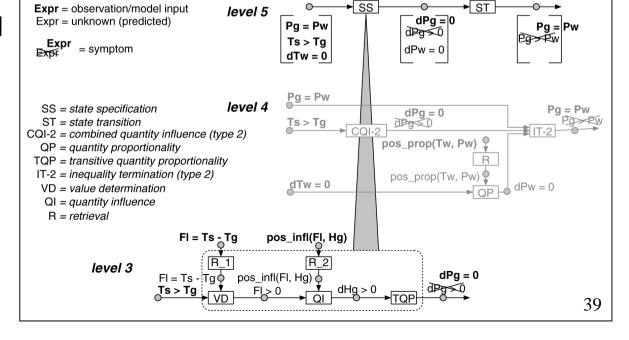
<u>Teacher</u>: What is the direction of the heat flow between source and gas?

- a. from source to gas
- b. from gas to source
- c. there is no heat flow between source and gas

Learner: a. from source to gas.

(NB: correct answer)

• Diagnosis: [R2], [QI], and [TQP]



An Example Diagnosis (6) (probe 4b)

- Three components can be tested: [R2], [QI], or [TQP]
- Probe selection:
 - probabilities: 1.43 for [R2]; 2.0 for [TQP]; 1.0 for [QI]
 - measure points: $pos_infl(Fl,Hg) \rightarrow | 1.43 (1 + 2) | = 1.57$
 - Hg > 0 → |(1.43 + 1) 2| = 0.43 (best split)

<u>Teacher</u>: Is the heat of the gas initially:

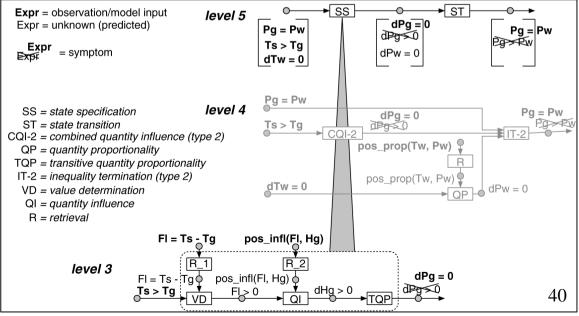
- a. increasing
- b. steady
- c. decreasing

Learner: a. increasing. (NB: correct answer)

- Diagnosis: single fault [TQP]
- Decompose: [TQP]

• Diagnosis:

4 single faults/candidates: [R3], [QP1], [R4] and [QP2]



An Example Diagnosis (7) (probe 5b)

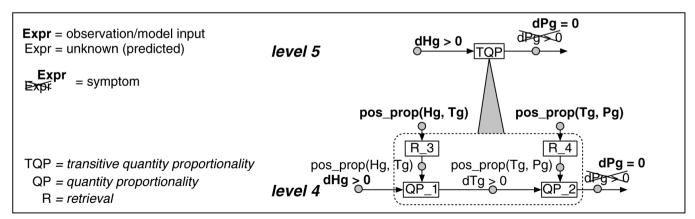
- Four components can be tested: [R3], [QP1], [R4] and [QP2]
- Probe selection:
 - probabilities: 1.43 for [R3] and [R4]; 1.0 for [QP1] and [QP2]
 - best measurement $\partial Tg > 0$ because |(1.43 + 1) (1.43 + 1)| = 0 (best split)

Teacher: Is the temperature of the gas initially:a. increasingb. steadyc. decreasingLearner: a. increasing.(NB: correct answer)

• Diagnosis:

2 single faults/candidates:

[R4] and [QP2]



An Example Diagnosis (8) (probe 6b)

- Two components can be tested: [R4] and [QP2]
- Probe selection:
 - probabilities: 1.43 for [R4]; 1.0 [QP2]
 - only one possible probe point *pos_prop*(Tg, Hg)

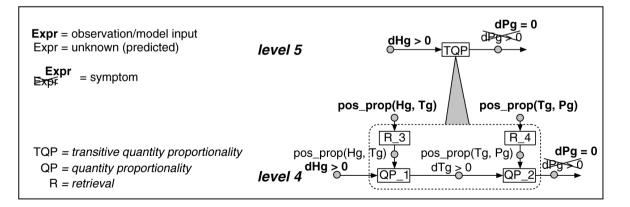
<u>Teacher</u>: What is the relation between the temperature

- and the pressure of the gas in the current state?
- a. if the temperature increases, then the pressure increases;
- b. if the temperature increases, then the pressure decreases;
- c. if the temperature increases, then this does not affect the pressure.

Learner: a. if the temperature increases, then the pressure increases. (NB: correct answer)

• Diagnosis: single fault [QP2]

(<u>meaning</u>: the learner does know the relation between temperature and pressure, but did not apply it here.)



GDE for Diagnosing Learner Behaviour

Evaluation with learners

STAR light		
volume left is equal to volume right level left is lower than level right width left is greater than width right volume left has become greater than volume right position has changed to right side up		
What will be the difference between water level left and water level right in the next state? a: water level left will be higher than water level right b: water level left will be equal to water level right c: water level left will be lower than water level right		
A B C Give me a hint	Quit	

On each side of the balance sits a container partially filled with water. The containers are equal in weight when empty, and have an equally sized outlet in the bottom. Through this outlet, the water flows out of the container, thereby decreasing the weight on that side of the balance. The flow rate of the two contained liquids can be different, corresponding to the pressure at the bottom. As a consequence, the balance moves to new positions, but the final state is always an equilibrium. 43

GDE for Diagnosing Learner Behaviour

Evaluation with learners

- 9 subjects: 4 'novices' and 5 'experts'
- 4 exercises per subject (about 30 minutes)
- 707 questions answered
- 30 diagnostic sessions
- average number of probes: less than 3
- often triggers self-repair
- explanation insufficient for novices

Main results:

- automatic generation of hierarchical articulate simulation models
- successful application of model-based diagnosis
- domain independent